Calculation of Electronic Structures and Magnetic Properties of C-quartz SiO₂ with and without Transition Metal Elements การคำนวณโครงสร้างอิเล็กทรอนิก และคุณสมบัติทางแม่เหล็ก ของอัลฟา-ควอทร์ซ ซิลิกาที่เจือด้วยโลหะแทรนซิชัน

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Abstract

Electronic structures and magnetic properties of pure α -quartz SiO₂ and doped α -quartz Si_{1-x} M_x O₂, where M represents the transition metals V, Cr and Mn, are calculated using the Gaussian98 program with the Hartree-Fock (HF) and Density Functional Theory (DFT) methods. In this calculation Si₁₈ O₂₆ H₃₂ and Si₁₇ MO₂₆H₃₂ represent pure and doped α -quartz SiO₂ respectively. The total energy calculated by DFT is found to be smaller than that calculated by HF and the lowest is found in Si₁₇ MnO₂₆H₃₂. The results show that the energy gaps of Si₁₈ O₂₆ H₃₂ calculated by HF and DFT methods are 15.766 eV and 7.560 eV respectively. The energy gaps of doped α -quartz SiO₂ calculated by both methods, are found to reduce and the narrowest one is found in Si₁₇ CrO₂₆ H₃₂. The calculations of the total dipole moments show that these values are higher in doped α -quartz SiO₂ than in pure α -quartz SiO₂, while the highest is found in Si₁₇ MnO₂₆ H₃₂.

บทคัดย่อ

งานวิจัยนี้ได้ทำการคำนวณโครงสร้างอิเล็กทรอนิก และสมบัติทางแม่เหล็กของอัลฟา-ควอทร์ซ ซิลิกา (α -quartz SiO₂) บริสุทธิ์และที่เจือด้วยโลหะแทรนซิชัน V, Cr และ Mn โดยใช้โปรแกรม Gaussian 98 ซึ่งได้ทำ การคำนวณโดยวิธี Hartree-Fock (HF) และวิธี Density Functional Theory (DFT) ในการคำนวณนี้ได้ใช้ Si₂O₂H₃₂ แทนอัลฟา-ควอทร์ซ ซิลิกาบริสุทธิ์ และ Si₁₇MO₂₆H₃₂ แทนอัลฟา-ควอทร์ซ ซิลิกาที่เจือด้วยโลหะแทรนซิชัน ผลที่ ได้จากการคำนวณพบว่าค่าพลังงานรวมที่ได้จากวิธี Density Functional Theory (DFT) จะมีค่าน้อยกว่าค่าพลังงาน รวมที่ได้จากกิรี Density Functional Theory (DFT) จะมีค่าน้อยกว่าค่าพลังงาน รวมที่ได้จากกิรี Hartree-Fock (HF) สำหรับทุก ๆ โครงสร้าง โดยโครงสร้างของ Si₁₇MO₂₆H₃₂ มีค่าพลังงานรวม ที่ได้จากกิรี Hartree-Fock (HF) สำหรับทุก ๆ โครงสร้าง โดยโครงสร้างของ Si₁₇MO₂₆H₃₂ มีค่าพลังงานรวม ที่ได้จากกิรี Hartree-Fock (HF) สำหรับทุก ๆ โครงสร้าง โดยโครงสร้างของ Si₁₇MO₂₆H₃₂ มีค่าพลังงานรวม ต่ำที่สุด นั่นคือ มีความเสถียรมากที่สุด และเมื่อวิเคราะห์หาค่า energy gap พบว่าค่า energy gap ของโครงสร้าง Si₁₇O₂₆H₃₂ ที่ได้จากการคำนวณโดยวิธี HF และ DFT มีค่าเท่ากับ 15.766 eV และ 7.560 eV ตามลำดับ แต่เมื่อ เจือด้วยโลหะแทรนซิชัน พบว่า ค่า energy gap ที่ได้จากการคำนวณทั้ง 2 วิธีมีค่าลดลง โดยโครงสร้าง Si₁₇CrO₂₆H₃₂ มีค่า energy gap น้อยที่สุด จากการคำนวณค่าไดโพลโมเมนต์รวม (Total dipole moment) พบว่าอัลฟา-ควอทร์ซ ซิลิกา ใจโพลโมเมนต์รวมเพิ่มขึ้นจากอัลฟา-ควอทร์ซ ซิลิกา ใดโพลโมเมนต์รวมเพิ่มขึ้นจากอัลฟา-ควอทร์ซ ซิลิกาบริสุทธิ์ โดย Si₁₇MnO₂₆H₃₂ มีค่า ไดโพลโมเมนต์รวมเพิ่มขึ้นจากอัลฟา-ควอทร์ซ ซิลิกาบริสุทธิ์ โดย Si₁₇MnO₂₆H₃₂ มีค่า

Keywords: Electronic structure, total dipole moment, α-quartz SiO₂ คำสำคัญ: โครงสร้างอิเล็กทรอนิก ผลรวมของค่าไดโพลโมเมนต์ อัลฟา-ควอทร์ซ ซิลิกา

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Introduction

Silica, SiO₂, in both amorphous and crystal forms is a very well-known material that is used worldwide due to its application in various fields such as cosmetics, drugs, foods, ceramic and glass industries, optical fibers, catalysis and microelectronics. General data and some details of silica's properties and its derivatives can be found in many textbooks in chemistry and materials science. Moreover, there is also a special monograph on silica (Devine et al., 2000) and a vast number of research articles. In general, silica is an insulator or semiconductor depending on its crystal structure and the width of the energy gaps. In addition, silica itself is not a ferromagnetic material. Recently, however, a study of its electronic structure and magnetic properties suggest that silica might be useful for spintronic (or spin-based electronics) applications (Wolf et al., 2001). In this technique, the spin of the electron carries information that can be used as an added degree of freedom in novel electronic devices. Thus, the development of functional ferromagnetic semiconductors is a key to the development of spintronic devices that will certainly be devices utilized in the future. Magnetic semiconductors based on non-magnetic semiconductors are so called dilute magnetic semiconductors (DMS).

There are some theoretical studies of α -quartz SiO₂ doped with some transition metals and non-transition metals which reveal that these materials can show ferromagnetic properties at room temperature (Dihn et al., 2005). In this work the electronic structures and total magnetic moments of pure α -quartz SiO₂ and doped α -quartz SiO₂ with Mn, V and Cr have been calculated using Hatree Fock (HF) and Density Functional Theory (DFT) methods (Jensen, 1988).

Structure of α -quartz SiO

The structure of α -quartz SiO₂ is described by the space group P3₂21 and its hexagonal primitive vectors are: $\dot{A}_1 = \frac{1}{2}a\hat{e}_x - \frac{1}{2}\sqrt{3}a\hat{e}_y$, $\bar{A}_2 = \frac{1}{2}a\hat{e}_x + \frac{1}{2}\sqrt{3}a\hat{e}_y$, $\bar{A}_3 = c\hat{e}_z$.

Table 1 shows the coordinates of Si and O in α -quartz SiO₂ structure consisting of 18 Si and 26 O. The structures of Si₁₈O₂₆ and Si₁₈O₂₆H₃₂ are shown in Figure 1(a) and Figure 1(b) respectively.

Material and Methods

The study of electronic structures and the magnetic properties of pure and doped α -quartz SiO₂ were based on ab initio calculations, Hartree–Fock (HF) and Density Functional Theory (DFT) methods. The Gaussian basis set used in these calculations was 6–31G* for both methods. According to this calculation, we used the structure of Si₁₈ O₂₆ H₃₂ as pure α -quartz and Si₁₇MO₂₆H₃₂ as doped α -quartz in which M were V, Cr and Mn. The calculation yielded the density of states (DOS) and the total dipole moment of each α -quartz SiO₂ structure. The GAUSSIAN98 package (Frisch et al., 1998) was used for all calculations. The flowchart of the calculation is shown in Figure 2.

Results and Discussion

Part 1 Calculation of the density of states and energy gap

Figures 3 to 6 illustrate the relationship between density of states and energy of different SiO_2 structures. Each figure depicts a plot of the density of state as a function of energy calculated by the Hatree Fock Method (HF). Figures 7 to 10 depict plots of the density of states as a function of the energy calculated by the Density Functional Theory Method (DFT).

From Figures 3 to 10, the energy gaps of pure α -quartz SiO₂ and doped α -quartz Si_{1-x} M_x O₂ can be calculated by the HF and DFT methods as summarized in Table 2.

From Table 2 it can be seen that the calculation by the DFT method yielded values of E_g smaller than those calculated by the HF method. The E_g of doped α -quartz SiO₂ was reduced in comparison to pure α -quartz SiO₂. This decrease in E_g should result from the creation of localized states due to transition metals in the gap.

Part 2 Calculation of the total dipole moments

Table 3 shows the values of total dipole moment for each SiO_2 structure calculated by the HF and DFT methods. These values could refer to the magnetic properties of each SiO₂ structure.

From Table 3 it can be seen that the total dipole moments of doped α -quartz Si_{1-x}M_x O₂ are higher than those for pure α -quartz SiO₂ except in the structure doped by V. Moreover, the highest value of total dipole moment is found in Si₁₇MnO₂₆H₃₂. From these results, it is expected that α -quartz SiO₂ could be ferromagnetic after doping with some transition metals.

Conclusions

We have used the HF and DFT methods to calculate the energy gaps, the density of states and the total dipole moments of α -quartz SiO₂ and doped α -quartz Si_{1-x}M_xO₂. The energy gaps of doped α -quartz Si_{1-x}M_xO₂ calculated by both HF and DFT methods are found to reduce as compared to the pure SiO₂ and the narrowest is found in Si₁₇CrO₂₆H₂₉.

The total dipole moments of doped α -quartz SiO₂ are higher than those of pure α -quartz SiO₂ except for doping with V. In addition, the highest value of total dipole moment is found in Si₁₇MnO₂₆H₃₂. The results confirm that it is possible to induce ferromagnetism in pure α -quartz SiO₂ doped with some transition metals such as Mn, V and Cr.

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316 Calculation of Electronic Structures and Magnetic Properties of $\alpha\text{-quartz SiO}_{_{\!\!\mathcal{P}}}$ with and without Transition Metal Elements

Atom	x	У	z	Atom	X	У	z	Atom	х	У	Z
Si	-3.7590	-1.9995	-1.8017	0	-2.7567	-2.4931	2.4435	0	-0.3000	-1.7620	2.9617
Si	-3.7590	1.9995	-3.6035	0	-3.8327	-3.1143	-0.6418	Si	3.6111	2.2556	-1.8017
0	-3.2374	-0.6212	-1.1600	0	-0.7807	-3.6339	1.1600	0	1.0807	3.1143	0.6418
Si	-3.7590	-1.9995	3.6035	Si	1.1544	-1.9995	-1.8017	0	2.1567	2.4931	-2.4435
Si	-3.7590	1.9995	1.8017	Si	1.1544	1.9995	-3.6035	0	2.1567	2.4931	2.9617
0	-3.2374	0.6212	1.1600	0	1.6760	-0.6212	-1.1600	Si	3.6111	-2.2556	-3.6035
Si	-1.3023	2.2556	-1.8017	0	-0.3000	1.7620	-2.9617	Si	3.6111	-2.2556	-3.6035
0	-0.7807	3.6339	-1.1600	Si	1.1544	-1.9995	3.6035	Si	3.6111	-2.2556	1.8017
Si	-1.3023	2.2556	3.6035	Si	1.1544	1.9995	1.8017	0	2.1567	-2.4931	-2.9617
0	-3.8327	3.1143	0.6418	Si	-2.3088	0.0000	0.0000	0	2.1567	-2.4931	2.4435
0	-2.7567	2.4931	-2.4435	0	-0.3000	1.7620	2.4435	0	1.0807	-3.1143	-0.6418
0	-2.7567	2.4931	2.9617	0	-1.3760	-1.1408	0.6418	Si	2.6046	0.0000	0.0000
Si	-1.3023	-2.2556	-3.6035	0	-1.3760	1.1408	-0.6418	0	3.5374	-1.1408	0.6418
0	-2.7567	-2.4931	-2.9617	0	-0.3000	-1.7620	-2.4435	0	3.5374	1.1408	-0.6418
Si	-1.3023	-2.2556	1.8017	0	1.6760	0.6212	1.1600				

Table 1. The coordinates of Si and O in $\alpha\text{-quartz SiO}_{_2}$ structure used in this study.

Table 2. The Energy gaps of pure and doped $\alpha\text{-quartz SiO}_{_2}$ calculated by HF and DFT methods.

Structure	Energy gap (eV), E _g					
Suucture	HF method	DFT method				
${\rm Si}_{18}{\rm O}_{26}{\rm H}_{32}$	15.766	7.560				
Si ₁₇ VO ₂₆ H ₃₂	4.359	1.760				
Si ₁₇ CrO ₂₆ H ₃₂	10.074	1.263				
Si ₁₇ MnO ₂₆ H ₃₂	10.062	3.091				

Starseture	Total dipole moment (Debye)					
Structure	HF method	DFT method				
${\rm Si}_{18}{\rm O}_{26}{\rm H}_{32}$	4.0479	3.6332				
$\rm Si_{17}VO_{26}H_{32}$	3.5726	2.7683				
Si ₁₇ CrO ₂₆ H ₃₂	4.7830	4.3398				
${\rm Si}_{17}{\rm MnO}_{26}{\rm H}_{32}$	5.1349	4.5009				

Table 3. The values of total dipole moments of each SiO_2 structure calculated by HF and DFT methods.



(a)

(b)

Figure 1. The structures of ${\rm Si}_{_{18}}{\rm O}_{_{26}}$ (a) and ${\rm Si}_{_{18}}{\rm O}_{_{26}}{\rm H}_{_{32}}$ (b).



Figure 2. Flowchart for the calculation of the density of states and the total dipole moment of pure and doped α -quartz SiO₂ based on HF and DFT methods (Kidkhunthod, 2006).



Figure 3. Plot of the density of state versus the energy of $Si_{18}O_{26}H_{32}$ structure using HF method.



Figure 4. Plot of the density of state versus the energy of $Si_{17}VO_{26}H_{32}$ structure using HF method. A is alpha state of electron and B is beta state of electron.





Figure 5. Plot of the density of states versus the energy of $Si_{17}CrO_{26}H_{32}$ structure using HF method.



Figure 6. Plot of the density of states versus the energy of $Si_{17}MnO_{26}H_{32}$ structure using HF method.



Figure 7. Plot of the density of states versus the energy of $Si_{18}O_{26}H_{32}$ structure using DFT method.



Figure 8. Plot of the density of states versus the energy of $Si_{17}VO_{26}H_{32}$ structure using DFT method.





Figure 9. Plot of the density of states versus the energy of $Si_{17}CrO_{26}H_{32}$ structure using DFT method.



Figure 10. Plot of the density of states versus the energy of $Si_{17}MnO_{26}H_{32}$ structure using DFT method.